Acta Crystallographica Section E Structure Reports Online ISSN 1600-5368 Editors: W. Clegg and D. G. Watson

A synchrotron redetermination of 2-(morpholinium-4yl)ethanesulfonate monohydrate, including a disordered water molecule

Maciej Kubicki, Dorota A. Adamiak, Wojciech R. Rypniewski and Agnieszka Olejniczak

Copyright © International Union of Crystallography

Author(s) of this paper may load this reprint on their own web site provided that this cover page is retained. Republication of this article or its storage in electronic databases or the like is not permitted without prior permission in writing from the IUCr.

Acta Cryst. (2007). E63, o2604-o2606

Kubicki *et al.* • $C_6H_{13}NO_4S \cdot H_2O$

organic papers

Acta Crystallographica Section E Structure Reports Online ISSN 1600-5368

Maciej Kubicki,^a* Dorota A. Adamiak,^b Wojciech R. Rypniewski^b and Agnieszka Olejniczak^c

^aDepartment of Chemistry, Adam Mickiewicz University, Grunwaldzka 6, 60-780 Poznań, Poland, ^bInstitute of Bioorganic Chemistry, Polish Academy of Sciences, Noskowskiego 12/14, 61-704 Poznań, Poland, and ^cCenter for Medical Biology, Laboratory of Medical Virology and Biological Chemistry, 93-232 Łódź, Poland

Correspondence e-mail: mkubicki@amu.edu.pl

Key indicators

Single-crystal synchrotron study T = 100 KMean σ (C–C) = 0.003 Å Disorder in main residue R factor = 0.045 wR factor = 0.127 Data-to-parameter ratio = 10.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A synchrotron redetermination of 2-(morpholinium-4-yl)ethanesulfonate monohydrate, including a disordered water molecule

The structure of the title compound, $C_6H_{13}NO_4S\cdot H_2O$ or $OC_4H_8NHCH_2CH_2SO_3\cdot H_2O$, was redetermined at 100 (1) K using synchrotron radiation. The water molecule, in contrast to the results of the prior room-temperature study [Christensen, Haxell, Lehmann & Nielsen (1993). *Acta Chem. Scand.* **47**, 753–756], was found to be disordered over two positions. The disordered model is more consistent with the observed hydrogen-bond network than the earlier suggested ordered structure.

Comment

The 2-(*N*-morpholino)ethanesulfonic acid (MES) buffer is commonly used in crystallization of proteins, nucleic acids and their derivatives. Crystals of MES monohydrate, (I), were obtained fortuitously during our crystallization trials of modified nucleosides (Olejniczak *et al.*, 2007).



The crystal structure determination of MES monohydrate reported earlier (Christensen *et al.*, 1993) is of relatively poor quality (room temperature, H atoms not refined, R > 7%). Most importantly, the solvent water molecule is described in that study as occupying a single position with relatively high displacement parameters. The results of our study at 100 K show that this water molecule can and, in fact, should be described as disordered over two positions. As the precise structural data might be important, for instance for modelling



Figure 1

The molecular structure of (I). Displacement ellipsoids are drawn at the 50% probability level; H atoms are depicted as spheres of arbitrary radius. Only the water molecule in the higher occupancy position is shown and the hydrogen bond is drawn as a dashed line.

 $\textcircled{\sc C}$ 2007 International Union of Crystallography All rights reserved

doi:10.1107/S1600536807018685

Received 21 February 2007 Accepted 15 April 2007

organic papers

1893 independent reflections

 $R_{\rm int} = 0.039$

1679 reflections with $I > 2\sigma(I)$

of



Figure 2

The two positions of the disordered water molecule and the hydrogen bonds (drawn as dashed lines). [Symmetry codes: (i) 2 - x, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) 2 - x, -y, -z; (iii) $x, -\frac{1}{2} - y, z + \frac{1}{2}$.]

purposes, we have decided to complete the full crystal structure determination.

The MES molecule exists in a zwitterionic form (Fig. 1). This is confirmed by successful refinement of atom H1 bonded to the morpholine atom N1, as well as by the approximately equal S-O bonds in the SO₃ group (Table 1). The bond length and angle pattern is similar to that found in related compounds (cf for instance Christensen et al., 1993; Deschamps et al., 2002; Milić et al., 2005); the morpholine ring is in a chair conformation [maximum values of the asymmetry parameters (Duax & Norton, 1975) are 4.7° for twofold axes and 5.9° for mirror planes]. The ethanesulfonate substituent occupies an equatorial position with respect to the morpholine ring $[C3-C2-N1-C11 = -179.7 (2)^{\circ}$ and C5-C6-N1- $C11 = -177.6 (2)^{\circ}$ and has an extended zigzag-like conformation $[N1-C11-C12-S12 = -171.0 (1)^{\circ}].$

The water molecule is disordered over two positions O1Wand O2W with site occupancy factors of 0.581 (6) and 0.419 (6), respectively. Interestingly, the water molecule in each of the two positions is involved in two similar hydrogen bonds, viz. as an acceptor for the morpholine N1/H1 group and as a donor to atom O122ⁱ. In both positions it forms one more bond as a donor; however, for one of the positions atom O123ⁱⁱ serves as an acceptor, whereas for the second position the role of the hydrogen-bond acceptor is played by atom O122ⁱⁱⁱ (Fig. 2; see the caption to Fig. 2 for the equivalent positions corresponding to the symmetry codes). As a result, one of the water positions gives rise to centrosymmetric dimers, whereas the alternative position produces chains along the [001] direction. One would thus expect the crystal structure to feature chain fragments which are arbitrarily interrupted by centrosymmetric dimers. Using graph-set notation (Etter et al., 1990; Bernstein et al., 1995), the latter can be described as $R_4^4(16)$, while the former is (second order) $C_2^2(8)$. Additionally, relatively strong C-H···O hydrogen bonds strengthen the structure (cf Table 2).

Experimental

The crystals were grown by the method of sitting-drop vapour diffusion (Weber, 1997). The reservoir solution contained 0.1 M

potassium chloride, 0.1 M magnesium chloride, 0.05 M MES buffer pH 6.0 and 10% w/v polyethyleneglycol (PEG 400) as precipitating agent. The solution used in the reservoir was from tube #9 of the Natrix crystallization kit purchased from Hampton Research (http:// www.hamptonresearch.com). The crystallization drop initially contained equal proportions of the above solution and a solution of 6-N-{5-[3-cobalt-bis(1,2-dicarbollide)-8-yl]-3-oxa-pentoxy}-2'-O-deoxyadenosine at 10 mg ml⁻¹. Small crystals appeared within two weeks. At first we thought that the crystals were of the nucleoside, but analysis of the crystals revealed that they contained the MES monohydrate.

Crystal data

$C_6H_{13}NO_4S \cdot H_2O$	V = 953.4 (3) Å ³
$M_r = 213.25$	Z = 4
Monoclinic, $P2_1/c$	Synchrotron radiation
a = 8.609 (1) Å	$\lambda = 0.8156 \text{ Å}$
b = 9.959 (2) Å	$\mu = 0.33 \text{ mm}^{-1}$
c = 11.144 (2) Å	T = 100 (1) K
$\beta = 93.79 \ (3)^{\circ}$	$0.4 \times 0.2 \times 0.2 \text{ mm}$

Data collection

MAR CCD 165mm diffractometer Absorption correction: none 6095 measured reflections

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$	H atoms treated by a mixture of
$wR(F^2) = 0.127$	independent and constrained
S = 1.06	refinement
1893 reflections	$\Delta \rho_{\rm max} = 0.31 \text{ e } \text{\AA}^{-3}$
180 parameters	$\Delta \rho_{\rm min} = -0.35 \ {\rm e} \ {\rm \AA}^{-3}$

Table 1

H	ĺyd	lrogen-	bond	geometry	(A,	0))
---	-----	---------	------	----------	-----	----	---

$D-\mathrm{H}\cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1-H1\cdotsO1W$	0.96 (3)	1.80 (3)	2.683 (3)	151 (2)
$N1 - H1 \cdots O2W$	0.96 (3)	1.83 (3)	2.783 (4)	175 (2)
$O1W - H1W2 \cdot \cdot \cdot O122^{i}$	0.85	1.98	2.790 (3)	159
$O2W - H2W1 \cdots O122^{i}$	0.85	1.97	2.727 (4)	147
$O1W-H1W1\cdots O123^{ii}$	0.85	1.98	2.775 (4)	156
$C2-H2A\cdots O123^{ii}$	0.97 (3)	2.52 (3)	3.366 (3)	145 (2)
$C11 - H11B \cdot \cdot \cdot O123^{ii}$	0.98 (3)	2.52 (3)	3.362 (3)	145 (2)
$O2W - H2W2 \cdot \cdot \cdot O122^{iii}$	0.85	2.01	2.834 (5)	162
$C12-H12A\cdots O4^{iv}$	0.96 (3)	2.52 (3)	3.351 (3)	145 (2)
$C6-H6B\cdots O121^{v}$	0.98 (2)	2.38 (2)	3.293 (3)	154.3 (17)

Symmetry codes: (i) -x + 2, $y + \frac{1}{2}$, $-z + \frac{1}{2}$; (ii) -x + 2, -y, -z; (iii) x, $-y - \frac{1}{2}$, $z + \frac{1}{2}$; (iv) $-x + 1, y - \frac{1}{2}, -z + \frac{1}{2}; (v) - x + 1, -y, -z.$

The water molecule was found in two alternative positions. Site occupation factors were refined at 0.581 (6) and 0.419 (6). O atoms were refined anisotropically without any constraints; the positions of water H atoms were calculated on the basis of the potential hydrogen-bond scheme with an O-H distance of 0.85 Å, and refined as riding with $U_{\rm iso}$ calculated as 1.2 times $U_{\rm eq}$ for the O atoms. All other H atoms were refined isotropically [N-H = 0.96 (3) Å; C-H0.94 (3)-1.06 (3) Å]. The refinement without splitting the water molecule was significantly worse; the R factor was 0.078 and the largest component of the displacement tensor of the O atom was as high as 0.5.

Data collection: MARCCD (MarResearch, 2002); cell refinement: DENZO/SCALEPACK (Otwinowski & Minor, 1997); data reduction: DENZO/SCALEPACK; program(s) used to solve structure:

organic papers

SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: Stereochemical Workstation Operation Manual (Siemens, 1989); software used to prepare material for publication: SHELXL97.

This work was supported by the European Community Research Infrastructure Action under the FP6 'Structuring the European Research Area Programme' (contract number RII3/CT/2004/5060008).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). Angew. Chem. Int. Ed. Engl. 34, 1555–1565.
- Christensen, A. N., Hazell, R. G., Lehmann, M. S. & Nielsen, M. (1993). Acta Chem. Scand. 47, 753–756.
- Deschamps, J. R., Flippen-Anderson, J. L. & George, C. (2002). Acta Cryst. E58, m167–m168.

Duax, W. L. & Norton, D. A. (1975). Atlas of Steroid Structure, Vol. 1, pp. 462-473. New York: Plenum.

Etter, M. C., MacDonald, J. C. & Bernstein, J. (1990). Acta Cryst. B46, 256–262. MarResearch (2002). MARCCD. MarResearch GmbH, Norderstedt, Germany.

- Milić, D., Řenić, M. & Matković-Čalogović, D. (2005). Acta Cryst. E61, m757– m758.
- Olejniczak, A. B., Plesek, J. & Lesnikowski, Z. J. (2007). Chem. Eur. J. 13, 311–318.
- Otwinowski, Z. & Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Siemens (1989). Stereochemical Workstation Operation Manual. Release 3.4. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Weber, P. C. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 13–22. New York: Academic Press.